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An efficient synthesis of optically pure (S)-2-functionalized 1,2,3,4-tetrahydroquinoline

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Abstract—Using a copper-catalyzed coupling reaction of amino acid and aryl halide, followed by intramolecular cyclization of N-aryl-1-hydroxyl-3-propylamines under the Swern's condition as the key steps, (S)-2-hydroxymethyl-1,2,3,4-tetrahydroquinoline was synthesized as an example of optically pure 2-functionalized 1,2,3,4-tetrahydroquinolines. © 2003 Elsevier Ltd. All rights reserved.

Substituted tetrahydroquinolines are the core structures in many important pharmacological agents and drug molecules, such as anti-arrhythmic and cardiovascular agents, anti-cancer drugs, immunosuppressants, ligands for 5-HT1A and NMDA receptors.¹⁻⁸ In fact 2-substituted tetrahydroquinoline oxamniquine (1) has been used in clinic to treat Manson's schistosomiasis since 1979. For these reasons, many efforts have been made to synthesize tetrahydroquinoline derivatives.^{9–12} Recently, we found that N-substituted 2,3,4,4a,5,6-hexahydro-1Hpyrazine[1,2-a]quinolines (2) are quite potent ligands for the dopamine 3 (D_3) receptor.¹³ Although the racemic form of 2 can be synthesized efficiently using previously published methods,^{14,15} the synthesis of the optically pure form of 2 has not been reported. The key intermediate to obtain the optically pure form of 2 is the chiral 2-hydroxymethyl tetrahydroquinoline 3. Herein, we report an efficient method to synthesize the optically pure substituted 2-hydroxymethyl tetrahydroquinoline 3 (Fig. 1).

Our initial idea was to use copper-catalyzed coupling reaction of amino acid and aryl halide, which was initially developed by Ma et al.¹⁶ and modified by Hayes

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Figure 1. Examples of important tetrahydroquinoline derivatives.

and co-workers,¹⁷ followed by intramolecular acylation as the key reactions to synthesize (S)-2-hydroxymethyl-7-methoxy-1,2,3,4-tetrahydroquinoline (Scheme 1). 3-Bromoanisole was treated with L-aspartic acid under Hayes' condition, followed by esterification to obtain N-aryl-aspartic diester 4 in a yield of 65% (two steps). Compound 4 was then reduced by NaBH₄, and the resulting diol 5 was reacted with triphosgene to produce oxazolidinone 6 and 7 (approximately 3:1 ratio). The major product 6 was oxidized by Jone's reagent, followed by intramolecular acylation to yield the desired cyclization product 9. Unfortunately, the yield of Jone's oxidation was quite poor (<15%), which might be due to the instability of the oxazolidinone under the very strong acidic condition, so a large quantity of 9 could not be obtained. Although the following intramolecular acylation, reduction and deprotection worked very well, an alternative method had to be developed to obtain 11 in satisfactory yield.

Since the major problem in Scheme 1 was the low yield of Jone's oxidation, we planned to produce the

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Scheme 2.

carboxylic acid 9 by another strategy. 4-(2-Hydroxy-ethyl)-3-(3-methoxy-phenyl)-oxazolidin-2-one 6 should be able to be oxidized under Swern's condition¹⁸ to yield the aldehyde 12 easily, and the aldehyde 12 could be oxidized to get the carboxylic acid 9 in a satisfactory yield (Scheme 2). Unexpectedly, under Swern's condition,¹⁸ the desired aldehyde 12 was not obtained. Instead, cyclization products 13, 14, 15 and 16 were produced, and the total yield was about 75% (Scheme 3, according to ¹H NMR, 13:14 = 10:1 and 15:16 = 10:1). This unexpected result might be attributed to the high reactivity of the disubstituted phenyl ring in 6. Compound 13 was purified by recrystallization (to remove the minor product 14) and the structure was determined by ${}^{1}H$ NMR, ¹³C NMR and X-ray crystallographic analyses (Fig. 2).

Both the reduction of double bond in compound 13 by hydrogenation and the removal of the methylthio group in 15 by Et_3SiH in trifluoroacetic acid produced compound 9 in excellent yields (Scheme 4). Finally, deprotection using the same reaction condition as shown in



Figure 2. X-ray structure of compound **13**. Thermal ellipsoids are at the 50% probability level, H atoms are shown as circles of an arbitrary radius.

Scheme 1 led to (S)-2-hydroxymethyl-tetrahydroquinoline, which can be used to synthesize optically pure





Scheme 4.

N-substituted 2,3,4,4a,5,6-hexahydro-1H-pyrazine[1,2-a]quinolines as a class of novel D₃ ligands.

In summary, using a copper-catalyzed coupling reaction of amino acid and aryl halide, followed by intramolecular cyclization of *N*-aryl-1-hydroxyl-3-propylamine derivatives under the Swern's condition as the key steps, a new and efficient method was developed to synthesize optically pure 2-hydroxymethyl tetrahydroquinoline. Further applications of this method for the synthesis of novel optically pure D_3 ligands and other compounds with interesting biological properties are underway in our laboratory and will be reported in due course.

References and notes

- Leeson, P. D.; Carling, R. W.; Moore, K. W.; Mosely, A. M.; Smith, J. D.; Stevenson, G.; Chan, T.; Baker, R.; Foster, A. C.; Grimwood, S.; Kemp, J. A.; Marshell, G. R.; Hoogsteen, K. J. Med. Chem. 1992, 35, 1954– 1968.
- Nagata, R.; Tanno, N.; Kodo, T.; Ae, N.; Yamaguchi, H.; Tamiki, N.; Antoku, F.; Tatsuno, T.; Kato, T.; Tanaka, Y.; Nakamura, M. J. Med. Chem. 1994, 37, 3956–3968.
- Guo, F.; Chang, B. H.; Rizzo, C. J. Bio. Med. Chem. Lett. 2002, 12, 151–154.

- 4. Nagata, R.; Tanno, N.; Kodo, T. PCT Int. Appl., WO 9308188, 1993.
- 5. Ejima, A.; Ohsuki, S.; Ohki, H.; Naito, H. U.S. 6,169,086.
- Stanton, J. L.; Ackerman, M. H. J. Med. Chem. 1983, 26, 986–989.
- Kojima, K.; Aizawa, Y.; Samata, N.; Sakai, J.; Koyama, K.; Tonohiro, T.; Sugimoto, M.; Hara, T.; Hisamoto, M.; Homma, H. PCT Int. Appl., WO 9623789, 1996.
- 8. Katritsky, A. R.; Rachwal, S.; Rachwal, B. *Tetrahedron* **1996**, *52*, 15031–15070, and references cited therein.
- 9. Gallou-Dagommer, I.; Gastaud, P.; Rajanbabu, T. V. Org. Lett. 2001, 3, 2053–2056, and references cited therein.
- Ma, D.; Xia, C.; Jiang, J.; Zhang, J.; Tang, W. J. Org. Chem. 2003, 68, 10442–10451.
- 11. Ma, D.; Xia, C.; Jiang, J.; Zhang, J. Org. Lett. 2001, 3, 2189.
- Fabio, R. D.; Alvaro, G.; Bertani, B.; Donati, D.; Giacobe, S.; Marchioro, C.; Palma, C.; Lynn, S. M. J. Org. Chem. 2002, 67, 7319–7328.
- Varady, J.; Wu, X.; Fang, X.; Min, J.; Hu, Z.; Levant, B.; Wang, S. J. Med. Chem. 2003, 46, 4377–4392.
- 14. Baxter, C. A. R.; Richards, H. C. J. Med. Chem. 1972, 15, 351.
- Rao, V. A.; Jain, P. C.; Anand, N. J. Med. Chem. 1970, 13, 516.
- Ma, D.; Zhang, Y.; Yao, J.; Wu, S.; Tao, F. J. Am. Chem. Soc. 1998, 120, 12459.
- Clement, J.-B.; Hayes, J. F.; Sheldrake, H. M.; Sheldrake, P. W.; Wells, A. S. *Synlett* 2001, 1423–1427.
- 18. Mancuso, A. J.; Swern, D. Synthesis 1981, 165-185.